Vibrational Investigation of DODC Cation for Recognition of Guanine Dimeric Hairpin Quadruplex Studied by Satellite Holes

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We have introduced the satellite hole spectral method to examine the binding sites of 3,3'-diethyloxadicarbocyanine cation (DODC⁺) to various guanine-rich oligonucleotides. The satellite hole patterns along with normal mode calculations allow us to determine the interaction of DODC⁺ with a dimeric hairpin quadruplex formed by sequence d(G₄T₄G₄). Our results are consistent with the groove binding model and eliminate the possibility of intercalation in the base pairs of guanine quartet.

I. Introduction

Telomeres, the end caps on eukaryotic chromosomes, protect chromosomal termini from degradation and fusion.¹⁻³ The telomeric DNA sequences generally consist of many tandem repeats of guanine-rich (G-rich) motifs, and various structures can be formed in a natural G-rich DNA sequences.⁴⁻⁵ Of particular interest is the quadruplex structures formed by the Hoogsteen hydrogen bond among four guanine bases.⁶ Recently, Shafer et al.⁷ have introduced the 3,3'-diethyloxadicarbocyanine iodide (DODCI) molecule to recognize the structure of guanine dimeric hairpin quadruplexes. The DODCI molecule is dissociated into I⁻ anion and DODC⁺ cation in protic solvents. The visible absorption spectrum of DODC⁺ has a maximum at 576 nm with a weak shoulder at 530-550 nm. Upon addition of dimeric hairpin quadruplex [d(G₄T₄G₄)]₂ (abbreviated as HQ), a new peak appears at 534 nm and a \sim 5 nm red shift of the 576 nm peak results. On the contrary, spectra similar to those of the free DODC⁺ spectrum were found for other structures with G-rich oligonucleotides, including the linear duplex [d(CGCGAATTCGCG)]₂ (LD) and the hairpin duplex d(CGCGT₄CGCG) (HD). The emergence of the 534 nm peak can be considered as a unique signature of the specific binding of the DODC⁺ to the dimeric hairpin quadruplex structure. Furthermore, they proposed that groove binding is more likely the binding mechanism based on the study of absorption, fluorescence, and CD spectra. However, the precise binding modes of the DODC⁺ to the dimeric hairpin quadruplexes remained uncertain. The purpose of this work is to determine the binding modes of DODC⁺ to the HQ.

Recently, we have studied satellite hole (SH) spectra of the BODIPY derivatives⁸ and the 9-aminoacridine^{9–11} for establishing the relationship between functional group and structural activity. The displacement of the SH from resonant hole provides the excited state frequency of the Franck–Condon active mode. More importantly, the intensity of the SH can be significantly enhanced when tuning the burning wavelength into the vibronic transition. This is because each SH has its own

electronic-vibrational transition at the burning wavelength, and therefore, it can be viewed as an individual electronic transition.¹⁰⁻¹¹ The high-resolution SH spectrum with a free background is very useful in monitoring the localized interactions of DNA-chromophore complexes.¹²⁻¹⁴ In this work, we introduce for the first time the SH spectrum, along with normal mode calculation, to identify binding modes of the DODC⁺ to the HQ. Furthermore, a binding structure of HQ:DODC⁺ is proposed.

II. Experimental Section

The experimental setup for the SH spectrum has been described elsewhere.¹⁵ Holes were produced by using a dye laser pumped by a mode-locked and Q-switched Nd:YAG laser (Quantronix 416).¹⁶ The laser emitted trains of ~10 pulses separated by 13 ns at a repetition rate of 500 Hz. The absorption spectra were obtained by dispersing the output of a xenon arc lamp (Oriel 66083) through a home-made double-beam spectrometer with a resolution of ~0.03 nm. Extreme care was taken to ensure that the cross section of the probe beam at the sample was inside the burning area.

DODCI was purchased from Exciton, Inc., and used without further purification. Oligonucleotides were purchased from Applied Biosystems. A stock solution of DODCI was prepared in the mixture of 5:4 glycerol:water (Gl:H₂O) solution. The ratio between dye and oligonucleotide in the Gl:H₂O solutions is 1:10 with the dye concentration of $\sim 3 \times 10^{-5}$ M and the buffer concentration of 0.01 M Tris-HCl (pH 7.5) and 0.15 M NaCl. Clear glass was normally formed by introducing the sample directly into a Janis dewar from room temperature to 6 K.

III. Results

Figure 1 shows the absorption spectra of the free DODC⁺ and the DODC⁺ mixed with each of the oligonucleotides of HD, LD, and HQ doped in Gl:H₂O solution at room temperature. The absorption spectra of the free DODC⁺, HD:DODC⁺, and LD:DODC⁺ are very similar to the absorption spectra reported by Shafer et al.,⁷ which consist of a prominent band at ~580nm with a weak shoulder at ~545 nm. However, the distinct peak at 534 nm observed in their spectrum of HQ:DODC⁺ is replaced

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Figure 1. Absorption spectra of $DODC^+$ (a), HD: $DODC^+$ (b), LD: $DODC^+$ (c), and HQ: $DODC^+$ (d) dissolved in 5:4 glycerl:water solution at room temperature.



Figure 2. Absorption and SH spectra of DODC⁺ (a), HD:DODC⁺ (b), LD:DODC⁺ (c), and HQ:DODC⁺ (d) dissolved in 5:4 glycerl:water solution at $T_B \approx 6$ K. The burn wavelength is located at $\lambda_B \approx 566$ nm. Prominent satellite holes are labelled with excited-state vibrational frequencies. The intensity of each SH spectrum is multiplied by 10.

by an additional shoulder at \sim 535 nm in our spectrum. The difference between these absorption spectra of HQ:DODC⁺ may be due to different solution.

Figure 2, plots a–d, shows the preburn absorption spectra and the SH spectra of the free DODC⁺, HD:DODC⁺, LD: DODC⁺, and HQ:DODC⁺ doped in Gl:H₂O glasses at $T_B \approx 6$ K, respectively. Note that the SH spectrum is obtained from the difference between each preburn and postburn absorption spectra. The burning wavelength (λ_B) is ~565 nm with the pulse energy of ~2.5 μ J and with a burn time of 5 min for each SH spectrum. Figure 2 shows that the bandwidth of the absorption band of HQ:DODC⁺ is broader than that of HD: DODC⁺ and LD:DODC⁺ by ~30%. However, the normalized hole depths are slightly weaker in the spectra of HQ:DODC⁺ than in those of HD:DODC⁺ and LD:DODC⁺.

Figure 3 shows the normalized SH spectra, the SH spectrum normalized by preburn absorption spectrum, taken at $\lambda_B \sim 575$ nm with the same pulse energy of $\sim 2 \mu J$ and with a burn time of 5 min for each spectrum. No appreciable difference is found for the SH frequencies among the spectra of free DODC⁺, LD: DODC⁺, and HD:DODC⁺. However, the SH at 163 cm⁻¹ in



Figure 3. Normalized SH spectra of DODC⁺ (a), HD:DODC⁺ (b), LD:DODC⁺ (c), and HQ:DODC⁺ (d) doped in Gl:H₂O glasses at T_B ≈ 6 K. The burn wavelength is located at $\lambda_B \approx 576$ nm. Prominent satellite holes are labelled with excited-state vibrational frequencies.



Figure 4. Normalized SH spectra of HD:DODC⁺ (a), HQ:DODC⁺ (b), taken at $\lambda_B \approx 581$ nm and HD:DODC⁺ (c), and HQ:DODC⁺ (d), taken at $\lambda_B \approx 586$ nm doped in Gl:H₂O glasses at T_B ≈ 6 K. Prominent satellite holes are labelled with excited-state vibrational frequencies.

the spectrum of free DODC⁺ is blue-shifted to 167 cm^{-1} in the spectrum of HQ:DODC⁺. In addition, a weak SH at \sim 240 cm⁻¹ is observed in Figure 3, plots a-c, but cannot be identified in Figure 3, plot d. On the other hand, a weak SH at 249 cm^{-1} is definitely observed in the spectrum of HQ:DODC⁺ but cannot be identified in other spectra. Another feature is that the depths of the low-frequency SHs are weaker in the spectra of HQ: DODC⁺ than in those of HD:DODC⁺ and LD:DODC⁺. However, the depths of the 530, 620, and 673 cm^{-1} SHs are similar among them. The weaker hole depths of the lowfrequency SHs may be caused by the red shift of the absorption origin band of HQ:DODC⁺, while the burning wavelength is kept constant. Considering the \sim 5 nm red shift of the absorption band for HQ:DODC⁺, Figure 4, plots a-d, show the normalized SH spectra of HD:DODC^+ and HQ:DODC^+ taken at $\lambda_B \sim 580$ and 585 nm with the similar pulse energy of $\sim 2 \mu J$ and with a burn time of 5 min for each spectrum, respectively. Figures 3 and 4 show that the shift of the absorption band indeed affects the hole depth. However, the hole depths of the low frequency SHs of HQ:DODC⁺ are still weaker than that of HD:DODC⁺.



Figure 5. Normalized SH sectra of DODC (a), HD:DODC⁺ (b), LD: DODC⁺ (c), and HQ:DODC⁺ (d) doped in Gl:H₂O glasses at T_B \approx 6 K. The burn wavelength is located at $\lambda_B \approx 556$ nm. Prominent satellite holes are labelled with excited-state vibrational frequencies.

Figure 5 shows another set of normalized SH spectra taken at $\lambda_B \sim 555$ nm. The same pulse energy of $\sim 2 \mu J$ is used for each spectrum with the burn time of 5 min for each spectrum. Figure 5 shows no significant difference in these SH spectra, except the decrease of the hole depth upon interaction with HQ. In particular, the depth of the 1118 cm⁻¹ SH is further reduced in the spectrum of HQ:DODC⁺.

IV. Mode Assignments

The assignments of the SHs are essential in identifying the binding modes involved in the HQ:DODC⁺ interactions. In

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order to distinguish the ring-chromophore, ethyl group, and methine chain modes of DODCI, we have further studied the SH spectra of 3,3'-diethyloxacarbocyanine iodide (DOCI), 3,3'diethylthiadicarbocyanine iodide (DTDCI), and 3.3'-dipropyloxacarbocyanine iodide (DOCI-3) doped in Gl:H₂O glass. The mode frequencies higher than 500 cm⁻¹ are compared to the Raman bands of DODC⁺ assigned by Gustafson and coworkers.¹⁷ In addition, Raman spectra of other cyanine dyes are also consulted.¹⁸ At present, we are not able to find the low-frequency vibrational study of DODC+. AM1 semiempirical calculations in vacuo by a Hyperchem package were conducted for the qualitative assignment of vibrational modes. The conformational geometry of DODC⁺ is first optimized by the Polak-Ribiere algorithm and then applied to perform normal mode calculation. Restricted Hartree-Fock wave functions are used. The calculated diagrams of the 170, 564, 887, and 1164 cm^{-1} modes of DODC⁺ are shown in Figure 6.

Our SH spectra show that the 136 and 228 cm⁻¹ SHs of DOC⁺ differ from the 163 and 240 cm⁻¹ SHs of DODC⁺. Note that the 167 cm⁻¹ mode of 1,3,5-hexatriene was assigned to C-C-C deformation of the polymethine chain.¹⁹ In addition, it is known that some out-of-plane vibrations of the retinal chain appear below 500 cm⁻¹.²⁰ AM1 calculations suggest that the 170 and 243 cm⁻¹ modes are mainly attributed to methine chain motion. The 530 cm^{-1} SH of DODC⁺ is consistent with the $527 \text{ cm}^{-1} \text{ SH of DTDC}^+$, but differs from the 558 cm⁻¹ SH of DOC⁺. Sato et al. ¹⁸ have tentatively assigned the band near 550 cm^{-1} of some carbocyanine dyes to the bending vibration of methine chain. AM1 calculation shows that the 564 cm⁻¹ mode involves the coupling of ring and chain motions. The 885 and 1078 cm^{-1} SHs of DOC⁺ are very similar to the 880 and 1080 cm⁻¹ SHs of DODC⁺. AM1 calculations show that the 887 and 1085 cm^{-1} modes are due to ring vibrations. In addition, Gustafson et al.¹⁷ have assigned the 884 and 1088 cm⁻¹ Raman bands of DODC⁺ to ring vibrations. Furthermore, the approximately identical frequency at 1118 cm⁻¹ SH observed in the spectra of DOC^+ and $DODC^+$ infers that this mode is





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Figure 6. Schematic diagrams of the 170, 564, 887, and 1164 cm^{-1} modes of DODC⁺.

TABLE 1: Mode Assignments of Prominent Satellite Holes of \mathbf{DODC}^+

DODC ⁺	assignment
163, 240	methine chain
530	chain and ring coupling
880, 962, 1080	ring chromophore
1118	ethyl group

most likely due to end chromophore motion. However, Sato et al.¹⁸ have assigned the 1135 cm⁻¹ band of DTDC⁺ to the vibration associated with the ethyl group. In order to test this assignment, we have studied the SH spectrum of DOCI-3. The disappearance of the 1118 cm⁻¹ SH and the occurrence of a very weak SH at 1130 cm⁻¹ in the spectrum of DOCI-3 support the assignment of 1118 cm⁻¹ SH of DODC⁺ to the ethyl vibration. The mode assignments of these SHs of DODC⁺ are listed in Table 1.

V. Discussion

The additional shoulder at 535 nm, the significant red shift, and the largest width of the absorption band of HQ:DODC+ among these DNA:DODC⁺ complexes indicate that the DODC⁺ is more perturbed by the HQ interactions. Shafer et al.⁷ have suggested that structural differences among HD, LD, and HQ are responsible for the specific dye recognition. This is because the HD and LD have two different grooves, but HQ has three different grooves characterized by wide, medium, and narrow grooves.²¹ The broader bandwidth of the absorption band of HQ:DODC⁺ can be described by more environmental interactions resulting from the additional groove of HQ structure. In addition, various widths among these grooves are determined by different orientation of backbone phosphate groups.²² The red shift of the absorption band of HQ:DODC⁺ suggests that a specific groove of HQ can possibly better interact with DODC⁺ since the red shift of the $\pi - \pi^*$ transition band is usually attributed to an increase in environmental polarity.²³

Besides the different absorption spectral features of HO: DODC⁺, the vibrational spectra can provide the information of the coupling modes of DODC⁺ to the HQ. Very similar spectral features, including frequencies and relative hole depths, of SHs among the free DODC⁺, HD:DODC⁺, and LD:DODC⁺ suggest that there is no specific binding of DODC⁺ to the HD and LD. However, the frequency shift of the SH from 163 cm⁻¹ to 167 cm^{-1} and the different features at ~240 cm^{-1} SH upon interaction with HQ infer that specific coupling is involved for the chain moiety of DODC⁺ to the HQ. Considering the electrostatic interaction between the phosphate group of the specific groove and the positive charge delocalized on the conjugated chain moiety of DODC+, the chain motions of DODC⁺ can be more perturbed by the HQ interactions. It is very possible that the chain moiety of DODC⁺ is trapped in the specific groove of HQ. In addition, the depths of the lowfrequency SHs of HQ:DODC⁺ are weaker than those of HD: DODC⁺, but the depths of the 530 and 620 cm⁻¹ SHs are very similar between them, supporting the claim that the chain moiety of $DODC^+$ is a coupling site to the HQ.

Furthermore, the approximately identical hole depths at 530, 620, and 673 cm⁻¹ in the SH spectra of HD:DODC⁺, LD: DODC⁺, and HQ:DODC⁺ suggest that DODC⁺ cannot be located in the pocket of the quartet of HQ. This feature also eliminates the possibility of intercalation in the base pairs of guanine quartet. Our previous study of 9-aminoacridine indicated that the hole depth of 9-AA is significantly reduced to less than 1/10 upon intercalation in DNA.¹⁴ In addition, the

study of thionine shows similar reduction of the hole depth upon intercalation with calf thymus.²⁴ Our results support that the groove binding plays an important role between HQ and $DODC^+$ interaction.

The hole depths of the ring modes of HO are weaker than that of HD and LD, implying that the interaction of the ring chromophore of DODC⁺ to the HQ differs from that to the HD and LD. The decrease of the hole depth on the ring modes of DODC⁺ is consistent with the proposed binding of the DODC⁺ to the HO, with one end chromophore A (for simplicity, we mark the two end chromophores as A and B) projecting into the thymine loop region and stacking between the loop thymines and the terminal guanine quartet.⁷ Another interesting feature is the increased reduction in depth of the 1118 cm⁻¹ SH peak in Figure 5d, implying that the coupling of the ethyl group of $DODC^+$ to the HQ can be strong. It further suggests that the end ring B has less interaction with HQ. More experiments are necessary to examine how the ethyl group of DODC⁺ interacts with HQ and how the chain length of DODC⁺ affects the recognition behavior.

In summary, we have illustrated that by using the highresolution satellite holes, especially including the low-frequency satellite holes, along with the characteristics of particular vibrations of DODC⁺, we are able to identify the local interactions of various structures of G-rich DNA. Our SH results are consistent with groove binding model and eliminate the possibility of ring intercalation in the base pairs of guanine quartet. The SH spectroscopic finding points to an interesting direction, namely the microscopic study of the local interactions of chromophores to oligonucleotides. The study of the binding modes and binding structures for the recognition of G-quartet should be important for our better understanding of the telomeres.

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